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Post-It® Fax Note 7671	Date July pages /		
To Reas Selsonie Co Dept.	From Josep Huly		
Co./Dept.	Co. /		
Phone #	Phone 217-524-4034		
742-886-4071	Fax #		

March 21, 2002

Mr. Stan Komperda
Project Manager
Illinois Environmental Protection Agency
Bureau of Land
1021 North Grand Avenue East
Springfield, Illinois 62794-9276

Re:

Groundwater/Enclosed Space Inhalation Risk Evaluation

Lockformer Site, Lisle, Illinois

EPA Region 5 Records Ctr.



Dear Mr. Komperda:

In response to your recent comments, Parsons is pleased to provide the following summary of our evaluation of the inhalation risk posed by groundwater containing dissolved concentrations of trichloroethene (TCE) in the vicinity of the Lockformer site in Lisle, Illinois.

SUMMARY OF ANALYSIS

Parsons performed inhalation risk evaluation using the ASTM 1739-95 Risk-Based Corrective Action standard, Section X2.5, Ground Water – Inhalation of Enclosed-Space (Indoor) Vapors. This methodology was used to estimate the inhalation risk in the basement of a theoretical private residence located directly above a groundwater plume of TCE in a subsurface lithology consistent with that of the Lockformer site in Lisle, Illinois. Attachment A includes the relevant pages from the ASTM standard that were used in evaluating this risk. Attachment B contains the specific work sheets developed by Parsons, which include all of the assumed input parameters used in this evaluation and the referenced source for each parameter.

A detailed description of the calculation methodology is included below. In summary, the ASTM analysis indicates that at the maximum concentrations at which TCE has been detected in a private well in the vicinity of the Lockformer site (~20 ppb), the contribution to inhalation cancer risk is less than 1 x 10⁻⁶. Specifically, our analysis was performed using three assumed groundwater concentrations for TCE: 10 parts per billion (ppb), 50 ppb, and 1,000 ppb (or 1 part per million, ppm). The table below shows the resulting inhalation cancer risk posed by each of these assumed concentrations:

TCE Concentration (ug/l)	Inhalation Cancer Risk
10	2.03 x 10 ⁻⁷
50	1.02 x 10 ⁻⁶
1,000	2.03 x 10 ⁻⁵

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Environmental Policy & Science



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LIMITATIONS OF ANALYTICAL METHODOLOGY

Having supplied the results of our analysis above, we feel it is also important to point out the significant limitations of the analytical method described in the ASTM standard. A quick perusal of the attached worksheets shows the significant number of assumptions that need to be made in order to complete this analysis. The geometry of the lithology, the depth at which the TCE plume is traveling laterally, and the specific geometry of the foundation cracks through which TCE vapors are assumed to enter the indoor space all factor significantly into the results of this analysis; none of these input parameters to the analytical model are known with any degree of certainty for the Lockformer site.

For reasons explained in more detail in the following section, the analysis is particularly sensitive to the thickness of the capillary fringe layer, or in the particular ease of the Lockformer site, to the thickness of uncontaminated water that may exist above the horizon at which the TCE plume may be traveling laterally in bedrock. (This sensitivity is related to the fact that any thickness of uncontaminated groundwater will significantly inhibit the diffusion of TCE in an upward direction). An illustration of this model sensitivity is shown in the table below. Three different thicknesses of an uncontaminated, inhibiting groundwater layer were assumed. The corresponding TCE source concentration related to an inhalation risk of 10⁻⁶ was then calculated. As shown by the table below; the thickness of this groundwater layer is roughly proportional to the source TCE concentration corresponding to a risk level of 10⁻⁶.

Assumed Thickness of Inhibiting Groundwater Layer (ft)	TCE Concentration Corresponding to 10 ⁴ Inhalation Risk (ug/l)
1	10.6
5	49.2
10	97.5

Because the inhalation risk level varies so significantly with the variation of the inhibiting groundwater layer (an unknown parameter at the Lockformer site), it is important to view the results of our analysis through the context of this limitation. Still, the variation in risk level varies primarily in a conservative manner; i.e., the risk level is likely less than 10^{-6} in the vicinity of the Lockformer site, given all of the currently available information.

DETAILED DISCUSSION OF ANALYSIS

The chemical characteristics of TCE used in the analysis were obtained from Part 742, Illinois Administrative Code. Most of remaining input parameters were obtained directly from the ASTM standard. The inhalation cancer slope factor for TCE was provided by the IEPA.

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The site-specific parameters used in the calculations are depth to groundwater, thickness of the capillary fringe layer, and the thickness of the vadose zone. In our analysis, the air and water volumetric content of the pore space within the capillary fringe layer were modified to reflect the most likely transport mechanisms of the TCE plume at the Lockformer site.

Figure 1 in Attachment A shows modeling assumptions regarding the definition of depth to groundwater, thickness of the capillary fringe, and the thickness of the vadose zone. It is assumed in this model that the constant source of dissolved contamination is already present at the top of the water table and that no diffusion transport is needed for contamination to reach the top of groundwater table from this constant source.

Transport characteristics through soil and the capillary fringe zone depend on the thickness of the zone, the air and water volumetric content of the pore space within the zone, and the compound diffusivity through air and water. The effective diffusion coefficient is a measure for the combined effect of these factors. The air diffusivity coefficient for TCE is several orders of magnitude higher than the corresponding water diffusivity coefficient.

For that reason, the effective diffusion through the capillary fringe zone (containing mostly water) is significantly lower than the effective diffusion coefficient for the vadose zone (containing mostly air). The resulting overall effective diffusion coefficient (calculated for the entire zone over which diffusion takes place) depends most significantly on the thickness of the layer with the smallest diffusion coefficient (the capillary fringe layer). Accordingly, the thickness of this capillary fringe layer is a much more significant input parameter than the thickness of the vadose zone through which the TCE must diffuse.

For the purposes of this analysis, Parsons assumed that the TCE plume has traveled laterally through a network of bedrock fractures to the off-site residential neighborhood, and that at least some of the groundwater in the saturated zone above the bedrock (and beneath the private residences) has not been affected (as would be the case if the release of TCE had originated from directly above). This conclusion has yet to be proven with actual data, but is a reasonable assumption given the likely transport mechanisms of the off-site TCE plume.

The attached evaluation assumes that the constant source of dissolved contamination in bedrock is approximately 5 feet below the groundwater surface. Based on currently available data, this is a conservative assumption (i.e., the thickness of uncontaminated groundwater may be more than 5 feet). Parsons treated this 5-foot layer as a capillary fringe layer by adjusting the volumetric content of soil vapor for this layer to zero to reflect the fact that the entire 5-foot thickness is completely saturated with water.

SUMMARY OF ANALYSIS

Overall, our analysis of the available data using the best available models leads us to conclude that TCE groundwater concentrations above 50 ppb at the Lockformer site could potentially contribute to an inhalation cancer risk greater than 1 x 10⁻⁶; however, the highest groundwater TCE concentration actually observed in the vicinity of the Lockformer site is less

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than this level (~20 ppb). It should also be noted that the limitations of the calculation methodology should not be ignored; the results of this analysis are very sensitive to changes in input parameters, and our conclusion should only be viewed as a preliminary conclusion based upon the available data. The only way to confidently and quantifiably determine the inhalation risk in the private residences would be through a systematic and empirical air sampling program in the vicinity of the Lockformer site.

We appreciate the opportunity to provide you with this analysis. Please call Mr. Sasa Jazic at any time if you have questions related to this letter, or should require any other additional assistance.

Sincerely,

PARSONS CORPORATION

Sasa Jazic

Project Engineer

Richard M. Frendt, P.E.

Technical Director

SJ/RF:ko enclosures

cc:

Stan Black, IEPA
Maggie Carson, IEPA
Tracy Hurley, IEPA
Michelle Ryan, IEPA
Kendra Pohn, AGO
Howard Chinn, AGO

File: 739542

ATTACHMENT A ASTM 1739-95 Ground Water – Inhalation of Enclosed-Space (Indoor) Vapors

tions and parameters used to prepare the example look-up Table X2.1. The basis for each of these equations is discussed in X2.2 through X2.10.

X2.2 Air—Inhalation of Vapors (Outdoors/Indoors)—In this case chemical intake results from the inhalation of vapors. It is assumed that vapor concentrations remain constant over the duration of exposure, and all inhaled chemicals are absorbed. Equations appearing in Tables X2.2 and X2.3 for estimating RBSLs for vapor concentrations in the breathing zone follow guidance given in Ref (26). Should the calculated RBSL exceed the saturated vapor concentration for any individual component, "> P_{out} " is entered in the table to indicate that the selected risk level or hazard quotient cannot be reached or exceeded for that compound

and the specified exposure scenario.

X2.3 Ground Water—Ingestion of Ground Water—In this case chemical intake results from ingestion of ground water. It is assumed that the dissolved hydrocarbon concentrations remain constant over the duration of exposure. Equations appearing in Tables X2.2 and X2.3 for estimating RBSLs for drinking water concentrations follow guidance given in Ref (26) for ingestion of chemicals in drinking water. Should the calculated RBSL exceed the pure component solubility for any individual component, ">S" is entered in the table to indicate that the selected risk level or hazard quotient cannot be reached or exceeded for that compound and the specified exposure scenario (unless free-phase product is mixed with the ingested water).

TABLE X2.2 Equations Used to Develop Example Tier 1 Risk-Based Screening Level (RBSLs) Appearing in "Look-Up" Table X2.1—

Carolnagenic Effects **

MareCan To	ables X2.4 through X2.7 for definition of paramete	Carcinagenic Effects ^A
Medium	Exposure Route	Risk-Based Screening Level (RBSL)
Air	inheletion *	$RBSL_{av} \left[\frac{\mu g}{m^3 - 4r} \right] = \frac{7R \times BW \times AT_a \times 365 \frac{days}{years} \times 10^3 \frac{\mu g}{mg}}{SF_a \times R_{av} \times EF \times ED}$
Ground water	ingestion (potable ground water supply only)*	RBSL_ [TR × BW × AT _e × 365 days years [LH ₂ O] = SF _e × IR_, × EF × ED
Ground weter ^c	endosed-space (indoor) vapor inhelation®	$ABSL_{\infty} \left[\frac{mg}{(L+L_{0})} \right] = \frac{ABSL_{\infty} \left[\frac{\mu g}{m^{2} \cdot ar} \right]}{VF_{\max}} \times 10^{-2} \frac{mg}{\mu g}$
Ground water ^c	amblent (outdoor) vapor inhalation ^o	$ABSL_{\infty} \left[\frac{mg}{L+H_{0}O} \right] = \frac{ABSL_{ab} \left[\frac{Mg}{m^{3} \cdot ab^{2}} \right]}{VF_{weaks}} \times 10^{-6} \frac{mg}{Mg}$
Buritolei soti	ingestion of soil, inhelation of vapors and particulates, and dermal contact. ⁸	$RBSL_{0} \left[\frac{MQ}{(kQ+c)R} \right] = \frac{TR \times BW \times AT_{0} \times 365 \frac{degr_{0}}{years}}{EF \times ED \left[\left(SF_{0} \times 10^{-9} \frac{kQ}{mQ} \times (R_{ad} \times RAF_{0} + SA \times M \times RAF_{d}) \right) + (SF_{i} \times R_{ab} \times (VF_{aa} + VF_{p})) \right]}$
P.		For striking and excessed soils (0 to 1 m)
Notartece soil ^a	emblent (outdoor) vapor inhalation ^o	$ABSL_{a} \left[\frac{mg}{\log eoff} \right] = \frac{ABSL_{aff}}{VF_{extre}} \times 10^{-3} \frac{mg}{\mu g}$
Maurice solo	enclosed space (Indoor) vapor inhelation. ⁶	$ABSL_{p}\left[\frac{mg}{\log polit}\right] = \frac{ABSL_{ab}\left[\frac{MQ}{m^{2}-abr}\right]}{VF_{exap}} \times 10^{-6} \frac{mQ}{MQ}$
Surface solo	leaching to ground water ^D	$RBSL_{v}\left[\frac{mg}{kg-coll}\right] = \frac{RBSL_{vv}\left[\frac{mg}{L+l_{v}O}\right]}{UF_{vv}}$

⁽Note that all RESE, values should be compared with thermodynamic partitioning limits, such as solubility levels, maximum vapor concentrations, and so forth. If a RBS the relevant partitioning limit, this is an indication that the selected risk or hazard level will never be reached or exceeded for that chemical and the selected research.

Bif Revening towis for these media based on other considerations (for example, assthetic, background levels, environmental resource protection, and so forth) can be shall with these equations by substituting the selected target level for RBSL, appearing in these equations.

These equations are based on Ref (28).
These equations simply define the "cross-media partitioning factors," VF_g and UF_{gree}

TABLE X2.4 Exposure Parameters Appearing in Tables X2.2 and X2.3

Parameters	Definitions, Units	Residential	Commercial/Industrial
AT _e	averaging time for carcinogens, years	70 years	70 years ^A
AT,	averaging time for noncarcinogens, years	30 years	25 years ^A
BW	adult body weight, kg	70 kg	70 kg^
ED	exposure duration, years	30 years	25 years^
EF	exposure frequency, days/years	350 days/year	250 days/year^
IR _{eed}	soli ingestion rate, mg/day	100 mg/day	50 mg/day^
R _{as} -Indoor	delly indoor inhelesion rate, m ² /day	15 m ⁵ /dey	20 m ⁸ /day ^A
Routdoor	delly outdoor inhetetion rate, m3/day	20 m³/day	20 maday^
R_	daily water ingestion rate, Liday	2 Liday	1 L/day ⁴
uF	leaching factor, (mg/L-H_O\/jmg/kg-soil)—see Table X2.5	chemical-specific	chemical-specific
w	soil to sidn adherence factor, molom ²	0.5	0.5
w,	dermal relative absorption factor, volatiles/PAHs	0.5/0.05	0.5/0.05*
RAF.	oral relative absorption factor	1.0	1.0
ROSE,	risk-based screening level for medis I, mg/kg-soil, mg/L-H ₂ O, or µg/m²-sir	chemical-, media-, and exposure route-specific	chemical-, media-, and exposure route-specific
9 0,	inhalation chronic reference dose, mg/kg-day	chemical-specific	chemical-apacific
RO,	oral chronic reference dose, mg/kg-day	chemical-specific	chemical-apacific
84	skin surface area, cm2/day	3160	31604
F,	inhelation cancer slope factor, (mg/kg-day)-1	chemical-specific	chemical-apacific
F .	oral cancer slope factor, (mg/kg-day)-1	chemical-specific	chemical-specific
THÓ CHT	target hezard quotient for individual constituents, unitiess	1,0	1.0
r A	target excess individual fletime cancer riek, unitiess	for example, 10 ⁻⁴ or 10 ⁻⁴	for example, 10 ⁻⁴ or 10 ⁻⁴
F,	voletitzation factor, (mg/m²-air)/(mg/kg-aoif) or (mg/m²-air)/(mg/ L-H ₂ O}—see Table X2.5	chemical- and media-specific	chemical- and media-specific

[^] See Ref (27).

surface,

X2.4.2.4 No loss of chemical as it diffuses towards ground surface (that is, no biodegradation), and

X2.4.2.5 Steady well-mixed atmospheric dispersion of the emanating vapors within the breathing zone as modeled by a "box model" for air dispersion.

X2.4.3 Should the calculated RBSL, exceed the pure component solubility for any individual component, ">S" is entered in the table to indicate that the selected risk level or hazard quotient cannot be reached or exceeded for that compound and the specified exposure scenario.

X2.5 Ground Water—Inhalation of Enclosed-Space (Indoor) Vapors:

X2.5.1 In this case chemical intake results from the inhalation of vapors in enclosed spaces. The chemical vapors originate from dissolved hydrocarbons in ground water located some distance below ground surface. Here the goal is to determine the dissolved hydrocarbon RBSL that corresponds to the target RBSL for vapors in the breathing zeroe, as given in Tables X2.2 and X2.3. If the selected target vapor concentration is some value other than the RBSL or inhalation (that is, odor threshold or ecological criterion), this value can be substituted for the RBSL and X2.3.

X2.5.2 A conceptual model for the transport of chemicals from ground water to indoor sir is depicted in Fig. X2.2. For simplicity, the relationship between enclosed-space air and dissolved ground water concentrations is represented in Tables X2.2 and X2.3 by the "volatilization factor" VF_{mesp} [(mg/m³-air)/(mg/L-H₂O)] defined in Table X2.5. It is based on the following assumptions:

X2.5.2.1 A constant dissolved chemical concentration in ground water,

X2.5.2.2 Equilibrium partitioning between dissolved chemicals in ground water and chemical vapors at the ground water table,

X2.5.2.3 Steady-state vapor- and liquid-phase diffusion

through the capillary fringe, vadose zone, and foundation cracks,

X2.5.2.4 No loss of chemical as it diffuses towards ground surface (that is, no biodegradation), and

X2.5.2.5 Steady, well-mixed atmospheric dispersion of the emanating vapors within the enclosed space, where the convective transport into the building through foundation cracks or openings is negligible in comparison with diffusive transport.

X2.5.3 Should the calculated RBSL_w exceed the pure component solubility for any individual component, ">S" is entered in the table to indicate that the selected risk level or hazard quotient cannot be reached or exceeded for that compound and the specified exposure scenario.

X2.6 Surficial Soils—Ingestion, Dermal Contact, and Vapor and Particulate Inhalation:

X2.6.1 In this case it is assumed that chemical intake results from a combination of intake routes, including: ingestion, dermal absorption, and inhalation of both particulates and vapors emanating from surficial soil.

X2.6.2 Equations used to estimate intake resulting from ingestion follow guidance given in Ref (26) for ingestion of chemicals in soil. For this route, it has been assumed that surficial soil chemical concentrations and intake rates remain constant over the exposure duration.

X2.6.3 Equations used to estimate intake resulting from dermal absorption follow guidance given in Ref (26) for dermal contact with chemicals in soil. For this route, it has been assumed that surficial soil chemical concentrations and absorption rates remain constant over the exposure duration.

X2.6.4 Equations used to estimate intake resulting from the inhalation of particulates follow guidance given in Ref (26) for inhalation of airborne chemicals. For this route, it has been assumed that surficial soil chemical concentrations, intake rates, and atmospheric particulate concentrations remain constant over the exposure duration.

X2.6.5 Equations used to estimate intake resulting from

[#] See Ref (28).

the inhalation of airborne chemicals resulting from the volatilization of chemicals from surficial soils follow guidance given in Ref (26) for inhalation of airborne chemicals.

X2.6.6 A conceptual model for the volatilization of chemicals from surficial soils to outdoor air is depicted in Fig. X2.3. For simplicity, the relationship between outdoor air and surficial soil concentrations is represented in Tables X2.2 and X2.3 by the "volatilization factor" VF_{xx} [(mg/m³-air)/(mg/kg-soil)] defined in Table X2.5. It is based on the following assumptions:

X2.6.6.1 Uniformly distributed chemical throughout the

depth 0-d (cm) below ground surface,

X2.6.6.2 Linear equilibrium partitioning within the soil matrix between sorbed, dissolved, and vapor phases, where

TABLE X2.5 Volatilization Factors (VF_I), Leaching Factor (LF_{ew}), and Effective Diffusion Coefficients (D_I^{ew})

Symbol	Cross-Media Route (or Definition)	Equation
VF	Ground water → enclosed-space vapors	$VF_{\text{uniform}} = \frac{H\left[\frac{D_{\text{eff}}/L_{\text{obs}}}{BR L_{0}}\right]}{1 + \left[\frac{D_{\text{eff}}/L_{\text{obs}}}{BR L_{0}}\right] + \left[\frac{D_{\text{eff}}/L_{\text{obs}}}{D_{\text{obs}}/L_{\text{obs}}}\right]} \times 10^{3} \frac{L}{\text{m}^{2}} \text{ A}$
VF _{warth}	Ground water → emblent (outdoor) vapors	$VF_{\text{waves}}\left[\frac{(mg/L+H_2O)}{(mg/L+H_2O)}\right] \approx \frac{H}{1 + \left[\frac{U_{\text{obs}}L_{\text{env}}}{WO_{\text{env}}^{\text{obs}}}\right]} \times 10^3 \frac{L}{m^3}$
VF _{se}	Surficial solis → ambient air (vapors)	$VF_{ac}\left[\frac{(mg/m^3-air)}{(mg/kg-aair)}\right] = \frac{2W_{\rho_0}}{U_{ac}k_{ac}} \sqrt{\frac{D_0^{an}H}{\sigma\{\delta_{ac}+k_{a}\rho_0+H\theta_{ac}\}\tau}} \times 10^3 \frac{\text{cm}^3\text{-kg}}{\text{m}^3\text{-g}} \text{c}$ or: $VF_{ac}\left[\frac{(mg/m^3-air)}{(mg/kg-aair)}\right] = \frac{W_{\rho_0}d}{U_{ac}k_{ac}\tau} \times 10^3 \frac{\text{cm}^3\text{-kg}}{m^3\text{-g}}; \text{ whichever is Jeas}^{\circ}$
VF,	Surficial solls → ambient air (perticulates)	$VF_{\mu} \left[\frac{(mg/m^2 - air)}{(mg/kg - air)} \right] = \frac{P_{\sigma}W}{U_{\sigma A_{ab}}} \times 10^{6} \frac{cm^{2} \cdot kg}{m^{2} \cdot g} e$
VF _{aarra}	Subsurface solls amblent air	$VF_{\text{numb}}\left[\frac{(\text{img/m2-olf})}{(\text{img/lig-colf})}\right] = \frac{H_{\theta_0}}{\left[\theta_{\text{no}} + K_{\text{ols}} + H\theta_{\text{no}}\right]\left(1 + \frac{U_{\text{cl}}k_{\text{ol}}L_{\theta}}{D_{\text{o}}^{\text{eff}}W}\right)} \times 10^3 \frac{\text{cm}^3 \cdot \text{kg}}{\text{m}^3 \cdot \text{g}}$
VF	Subsurface solf → enclosed-space vapors	$VF_{aaap} \left[\frac{(mg/m^2-ak^2)}{(mg/m^2-ac^2)} \right] = \frac{H\rho_0}{\{\theta_{aa} + k_{a}\rho_0 + M\theta_{ab}\}} \left[\frac{\Omega_0^{ad}/L_0}{ER L_0} \right] \times 10^9 \frac{cm^2 \log A}{m^2 - g} A$ $1 + \left[\frac{D_0^{ad}/L_0}{ER L_0} \right] + \left[\frac{D_0^{ad}/L_0}{(p_{abs}^2-ac^2)^2 + k_{a}^2 + k_{a}^2$
₽°**	Subsurface soits → ground water	$LF_{ab} = \frac{\left[\frac{\text{fmg/L-H_{a}O}}{\text{[fmg/Ncg-acid]}}\right] = \frac{\mu_{a}}{\left[\ell_{ma} + R_{a}\rho_{a} + H\theta_{ab}\right]\left(1 + \frac{U_{ab}A_{ab}}{M}\right)} \times 10^{a} \frac{\text{cm}^{2} \cdot \text{kg}}{\text{L-g}} \text{ a}$
D;ª	Effective diffusion coefficient in soil based on vapor-phase concentration	Der Cotte = Dar Gas + Dare H Gas A
Og.	Effective diffusion coefficient through foundation cracks	Det (ST) = Det (ST) + Det (H ST)
0#	Effective diffusion coefficient through capillary frings	0 m (m) = 0 m (m) + 0 m (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (m) (
D#	Effective diffusion coefficient between ground water and	0 = [
C	Soll concentration at which dissolved pore-water and vapor phases become seturated	$C_a^{\text{nef}}\left[\frac{mg}{kg\text{-coff}}\right] = \frac{S}{\rho_a} \times [IM_{aa} + \theta_{ava} + k_a\rho_a] \times 10^0 \frac{L\text{-g}}{\text{cm}^3\text{-kg}}^{\rho}$

⁴ See Ref (29). ⁶ See Ref (30).

^a See Ref (31).

^o Based on mess belance.

[#] See Ref (32).

[&]quot; See Ref (33).

TABLE X2.6 Soil, Building, Surface, and Subsurface Parameters Used in Generating Example Tier 1 RBSLs

	See X2.10 for justicization or parameter selection.		
a smoter	B Definitions, Units	Residential	Commercial/Industrial
ď	tower depth of surficiel soil zone, cm	100 di	100 pm
Dee.	diffusion coefficient in air, cm ^e /s	chemical-specific	chemical-specific
D	diffusion coefficient in water, cm²/s	chemical-apacitic	chemical-epecific
ER	enclosed-space air exchange rate, U/s	0.00014 a-1	0.00023 e ⁻¹
-	traction of organic carbon in soil, g-C/g-soil	0.01	0.01
Н	herey's law constant, (cm2-H ₂ O)/(cm2-eir)	chemical-specific	otyernical-apacific
n-	thickness of capillary fringe, on	5 cm	66m
h, "	thickness of vedose zone, om	295 cm	295 cm
ı	infitration rate of water through soil, onlyeers	30 cm/year	_30 cm/yeer
k	cerbon-water scrption coefficient, cm ⁴ -H ₂ O/g-C	chemical-specific	chemical-apacific
k,	sol-water sorption coefficient, cm2+H ₂ O/g-solt	$f_{aa} \times k_{aa}$	las × kas
L.	enclosed-space volume/infiltration area ratio, om	200 cm	300 cm
	enclosed-space foundation or well thickness, cm	15 cm	15 cm
-OW	depth to ground water = $h_{\rm eas} + h_{\rm pr}$ cm	300 cm	300 cm
4	depth to subsurface soil sources, on	100 cm	100 cm
P	particulate emission rate, g/cm²-e	6.9 × 10 ⁻¹⁴	6.9 × 10 ⁻¹⁴
s	pure component solubility in water, mg/L-H ₂ O	chemical-apacific	chemical-specific
J	wind speed above ground surface in ambient mixing zone, cm/s	225 om/a	225 om/s
J	ground weter Dercy velocity, onlywer	2500 cm/year	2500 cm/year
Ň.	width of source area parallel to wind, or ground water flow direction, on	1500 cm	1500 cm
-	ambient air mixing zone height, om	200 cm	200 cm
-	ground water mixing zone thickness, cm	200 cm	200 cm
	areal fraction of cracks in foundations/walls, cm2-cracks/cm2-total area	0.01 cm²-cracks/cm²-total area	0.01 cm²-cracks/cm²-total are
	volumetric air content in capitary tringe soils, cm ^e -air/cm ^e -soil	0.038 cm²-eir/cm²-eoli	0.38 cm ^s eir/om ^s eoil
-	volumetric air content in foundation/well cracks, cm2-air/cm3 total volume	0.26 cm ² eir/cm ² total volume	9.26 cins air/cm3 total võlume
	volumetric air contant in vadose zone soils, cm²-air/cm²-acil	0.26 cm²-air/cm²-eoil	0.26 cm ² -eir/cm ² -soil
	total soil porcetty, am ³ /cm ³ -soil	0.38 cm²/cm²-soli	0.36 cm²/cm²-so£
-	volumetric water content in capillary trings soils, cm ² -H ₂ O/cm ² -soil	0.342 cm ² -H _B O/cm ² -soli	0.342 cm²-H ₂ O/cm²-scii
	volumetric water content in foundation/wall cracks, cm ² -H _g O/cm ² total volume	0.12 cm ² H ₂ O/cm ² total volume	0.12 cm ² -H ₂ O/cm ² total volum
-	volumetric water content in vadose zone soils, cm4-H ₂ O/cm4-soil	0.12 cm ² /f _g O/cm ² -soli	0.12 cm²-H ₂ O/cm²-soit
	soil bulk density, g-eoil/cm ^a -soil	1.7 g/cm²	1.7 g/cm³
	everaging time for vapor tux, s	7.88 × 10 ⁴ s	7.86 × 10 ⁶ s

the partitioning is a function of constant chemical- and soil-specific parameters,

X2.6.6.3 Diffusion through the vadose zone,

X2.6.6.4 No loss of chemical as it diffuses towards ground surface (that is, no biodegradation), and

X2.6.6.5 Steady well-mixed atmospheric dispersion of the emanating vapors within the breathing zone as modeled by a "box model" for air dispersion.

X2.6.7 In the event that the time-averaged flux exceeds that which would occur if all chemical initially present in the surficial soil zone volatilized during the exposure period,

then the volatilization factor is determined from a mass balance assuming that all chemical initially present in the surficial soil zone volatilizes during the exposure period.

X2.7 Subsurface Soils-Inhalation of Outdoor Vapors:

X2.7.1 In this case chemical intake is a result of inhalation of outdoor vapors which originate from hydrocarbons contained in subsurface soils located some distance below ground surface. Here the goal is to determine the RBSL for subsurface soils that corresponds to the target RBSL for outdoor vapors in the breathing zone, as given in X2.2. If the selected target vapor concentration is some value other than

TABLE X2.7 Chemical-Specific Properties Used in the Derivation Example Tier 1 RBSLs

	17000	Other Park	openio i i openio				
Chemical	CAS Number	M., g/mal	H, LHLO/L-#	Der, ome/s	Orr, cm²/s	tog(Kas). L/kg	log(K _{ees}), L/k
Benzene	71-43-2	78^	0.22^	0.0934	1.1 × 10-64	1.654	2.13^
Tokene	106-86-3	924	0.26^	0.086^	9.4 × 10 ⁻⁶⁰	2.134	2.65^
	100-41-4	106^	0.32^	0.076^	8.5 × 10-40	1.984	3.13^
Ethyl benzene		106^	0.29^	0.0720	8.5 × 10-60	2.38^	3.26^
Mibrael stylenes	1330-20-7	128^	0.049^	0.0720	9.4 × 10-44	3,114	3.28^
Naphthelene Benzo(sjpyrene	91-20-3 50-32-8	252°	5.8 × 10-4 #	0.0600	5.6 × 10-4 P	5.59#	5.96*
Chemical	CAS Nurt	rber	SF, kg-day/mg	SF, kg-day/mg	<i>AID</i> ., m	g/kg-day	RID, mg/kg-day
Benzane	71-43-2		0.029"	0.029"	···		
Tokume	106-68-	3	•••	• • •	0.2		0.11
Ethyl benzene	100-41-	4			0.1		0.29*
Mixed xylenes	1330-20				2.0		2.0° 0.004°
Nachthalerie 91-20-3					0.0	0040	0.0045
Benaplebyrene			7.3°	8.1°	••		

⁴ See Ref (34).

See Ref (36).
 See Ref (7).

Diffusion coefficient calculated using the method of Fuller, Schettler, and Gliddings, from Ref (11).

^{*} Calculated from $K_{\rm ext}/K_{\rm ext}$ correlation: $\log(K_{\rm ext}) = 0.937 \log(K_{\rm ext}) = 0.006$, from Ref (11).

[&]quot; See Ref (2).

G See Ref (3).

the RBSL for inhalation (that is, odor threshold or ecological criterion), this value can be substituted for the $RBSL_{air}$ parameter appearing in the equations given in Tables X2.2 and X2.3.

X2.7.2 A conceptual model for the transport of chemicals from subsurface soils to ambient air is depicted in Fig. X2.4. For simplicity, the relationship between outdoor air and soil concentrations is represented in Tables X2.2 and X2.3 by the "volatilization factor," VF_{somb} [(mg/m³-air)/(kg-soil)], defined in Table X2.5. It is based on the following assumptions:

X2.7.2.1 A constant chemical concentration in subsurface soils.

X2.7.2.2 Linear equilibrium partitioning within the soil matrix between sorbed, dissolved, and vapor phases, where the partitioning is a function of constant chemical- and soil-specific parameters,

X2.7.2.3 Steady-state vapor- and liquid-phase diffusion through the vadose zone to ground surface.

X2.7.2.4 No loss of chemical as it diffuses towards ground surface (that is, no biodegradation), and

X2.7.2.5 Steady well-mixed atmospheric dispersion of the emanating vapors within the breathing zone as modeled by a "box model" for air dispersion.

X2.7.3 Should the calculated RBSL_s exceed the value for which the equilibrated vapor and dissolved pore-water phases become saturated, C_j^{sec} [mg/kg-soil] (see Table X2.5 for calculation of this value), "RES" is entered in the table to indicate that the selected risk level or hazard quotient cannot be reached or exceeded for that compound and the specified exposure scenario (even if free-phase product or precipitate is present in the soil).

X2.8 Subsurface Soils—Inhalation of Enclosed-Space (Indoor) Vapors:

X2.8.1 In this case chemical intake is a result of inhalation of enclosed-space vapors which originate from hydrocarbons contained in subsurface soils located some distance below ground surface. Here the goal is to determine the RBSL for subsurface soils that corresponds to the target RBSL for indoor vapors, as given in Tables X2.2 and X2.3. If the selected target vapor concentration is some value other than the RBSL for inhalation (that is, odor threshold or

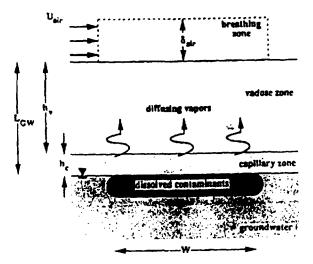
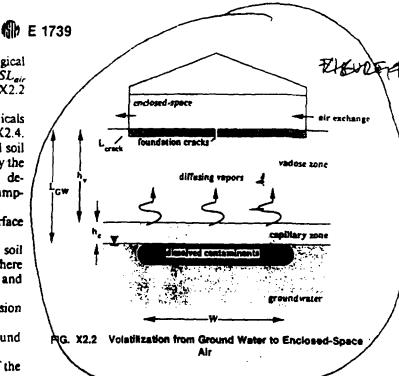


FIG. X2.1 Volatilization from Ground Water to Ambient Air



ecological criterion), this value can be substituted for the RBSL_{air} parameter appearing in the equations given in Tables X2.2 and X2.3.

X2.8.2 A conceptual model for the transport of chemicals from subsurface soils to enclosed spaces is depicted in Fig. X2.5. For simplicity, the relationship between indoor air and soil concentrations is represented in Tables X2.2 and $X\bar{2}.3$ by the "volatilization factor," VF_{sup} [(mg/m³-air)/(kg-soil)], defined in Table X2.5. It is based on the following assumptions:

X2.8.2.1 A constant chemical concentration in subsurface soils,

X2.8.2.2 Linear equilibrium partitioning within the soil matrix between sorbed, dissolved, and vapor phases, where the partitioning is a function of constant chemical- and soil-specific parameters,

X2.8.2.3 Steady-state vapor- and liquid-phase diffusion through the vadose zone and foundation cracks,

X2.8.2.4 No loss of chemical as it diffuses towards ground surface (that is, no biodegradation), and

X2.8.2.5 Well-mixed atmospheric dispersion of the emanating vapors within the enclosed space.

X2.8.3 Should the calculated RBSL_u exceed the value C_j^{rost} [mg/kg-soil] for which the equilibrated vapor and dissolved pore-water phases become saturated (see Table X2.5 for calculation of this value), "RES" is entered in the table to indicate that the selected risk level or hazard

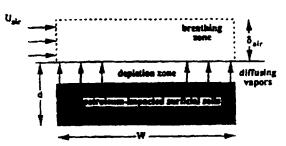


FIG. X2.3 Voletilization from Surficial Soils

ATTACHMENT B CALCULATION WORKSHEETS

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Private Residence Vicinity of Lockformer Facility Lisie, Illinois

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Chemical Compound: Trichloroethylene

			MPUT PARAMETERS	
			centrio-Specific Parameters	
Parameter	Value	Unite.	Description	Bourt
Scenario	Résidential	•	Scenario Type	·
TR	1E-06	unitiess	Torget Center Risk	AST
8W	70	(hg)	Adult Bady Weight	AST
ATC	70	(years)	Averaging Time for Cercinogene	AST
Rair-Indoor	15	(m3/d)	Delly Indoor Inhelation Rate	AST
EF	360	(4)17)	Exposure Frequency	AST
ED	30	(71)	Exposure Duration	AST
	- Material T		repound-Specific Parameters	
Parameter	Value	Unite	Description	Sour
Compound	Trichloroethylene	1	Name of Chemical Compound	 _
SFI	4.00E-01	(mg/tig-d)-1	Inheletion Cencer Stope Factor	IEP/
H	0.42	(unifices)	Henry's Law constant	TAC
Dair	0.079	(cm2/s)	Diffusion Coefficient in Air	TAC
Dwater	9.1E-06	(cm2/s)	Diffusion Coefficient in Weter	TAC
			Site-Specific Parameters	
Parameter	Value	Units	Describition	Sour
ni	0.01	(unitiess)	Areal Fraction of Cracks in Foundations/Walls	AST
ER	0.00014	(8-1)	Enclosed-Space Air Exchange Rate	AST
Law	1525	(cm)	Depth to Groundwester	- 8
Lb	200	(cm)	Enclosed-Space Volume/Infiltration Area Ratio	AST
Loreck	15	(OTT)	Enclosed-Space Foundation or Wall Thickness	AST
hcep	182	(OM)	Thickness of Capitary Prings	8
hv	1373	(cm)	Thickness of Vadose Zone	8
THYS	0.12	(unitions)	Volumetric Water Content in Vadore Zone Soile	AST
nes	0.26	(unitiese)	Volumetric Air Consent in Vadoes Zone Solls	AST
n	0.36	(unitiess)	Total Soil Porosity	AST
писар	0.36	(unitiess)	Volumetric Weter Content in Capitary Fringe Soils	8
песер	0	(unitiess)	Volumetric Air Content in Capitary Fringe Soils	3
rwcrack	0.12	(unitions)	Volumetric Water Content in Foundation/Wall Cracks	AST
necreck	0.26	(unitiese)	Volumetric Air Content in FoundationWall Cracks	AST

S	Site Specific Parameter
ASTM	ASTM Standard E 1730-05

TACO Table E, Default Physics//Chemical Parameters, Part 742, June 1998

IEPA Value provided by the IEPA

Eq Value calculated by previous equation

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Private Residence Vicinity of Lockformer Facility Lisle, Illinois

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Chemical Compound: Trichloroethylene

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		c	ALCULATED PARAMETERS	
~	Effective (Official	nt in Soil Based on Vaper-Phase Concentration (Ds-eff)	
Input Parameters	Value	Units	Description	Source
Dair	0.079	(cm2fs)	Diffusion Coefficient in Air	TACO
Dwater	9.10E-08	(cm2/s)	Diffusion Coefficient in Water	TACO
H*	0.422	(unitiess)	Henry's Law constant	TACO
Thes	0 12	(unitiess)	Volumetric Water Content in Vadose Zone Soits	ASTM
nes .	0.26	(unitiess)	Volumetric Air Content in Vadose Zone Solis	ASTM
n	0.36	(unitions)	Total Soil Porcelly	ASTM
Calculated Parameter	Value	Units :	Cestription	Source
De-eff	6.14E-63	(cm2/s)	Effective Diffusion Coefficient in Soil	Eq.
Innet Presentation	Visionia .	Unite	Desire Manual Control of the Control	Source
			ifficient through Foundation Create (Dareak-eff)	
Ingut Parameters	Value	Units	Deber billigen (Source
Deir	0,078	(cm2/s)	Diffesion Coefficient in Air	TACO
Dweler	9.100000000E-06	(cm2/s)	Diffusion Coefficient in Water	TACO
HP	0.422	(unitiess)	Henry's Law constant	TACO
nwcrack	0.12	(unitiess)	Volumetric Water Content in Foundation/Wall Cracks	ASTM
necneck	0.26	(unifices)	Volumetric Air Confert in Foundation/Wall Cracks	ASTM
n	0.36	(unitiess)	Total Soil Porcelly	ASTM
Catastated Parameter	Value	Unite	Description .	Source
Derack-off	8.14E-83	(cm2/a)	Effective Ottopion Coefficient through Foundation Creeks	Eq.
	2	ffective Diffusion (Coefficient through Capillary Pringe (Dcap-eff)	
Input Paremeters	Value	Units	Description:	Source
Detr	0.079	(cm2/s)	Offusion Coefficient in Air	TACO
Dwater	9.105-06	(cm2/s)	Diffusion Coefficient in Water	TACO
H	0.422	(unificee)	Herry's Law constant	TACO
nwcep	0.36	(unitiess)	Volumetric Water Content in Capitary Fringe Soils	S
necep	0	(unitiess)	Volumetric Air Content in Capitary Fringe Soile	8
ħ	0.38	(unitiess)	Total Soil Porcelty	ASTM
				
Calculated Parameter	Value	Units	Description .	Source

8	Site Specific Perantianar
ASTM	ASTM Standard E 1739-95
TACO	Table E, Default Physical/Chemical Parameters, Part 742, June 19
IEPA	Value provided by the IEPA

Value calculated by previous equation

Eq

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Private Residence Vicinity of Lockformer Facility Lisle, Illinois

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Chemical Compound: Trichloroethylene

		CA	LCULATED PARAMETERS	
	Effect	ive Diffusion Coefficie	nt between Groundwater and Boll Surface (Dws-eff)	
Input Parameters	Value	Units	Description	Source
hcap	152.4	(cm)	Thickness of Capitary Fringe	S
hv	1372.6	(cm)	Thickness of Vadose Zone	S
Ocsp-eff	5.94E-06	(cm2/s)	Effective Diffusion Coefficient through Capitary Fringe	Eq.
Ds-eff	6.14E-03	(cm2/s)	Effective Diffusion Coefficient in Soil	Eq.
siculated Parameter	Value	Units	Description	Source
Dws-eff	\$.00E-06	(cm2/s)	Effective Diffusion Coefficient between Groundwater and Solf Burface	Eq.
Input Parameters	Value	Unite	Description	Source
H H	0.422	(unitiess)	Henry's Litw constant	TACO
Dws-eff	5.80E-06	(am2fe)	Effective Diffusion Coefficient between Groundwater and Soil Surface	Eq.
Law	1525	(cm)	Depth to Groundwater	8
ER	0.00014	(0-1)	Endicaci-Space Air Exchange Rate	ASTM
Lb	200	(cm)	Enclosed-Space Volume/infiltration Area Ratio	ASTM
Dorack-eff	6.14E-03	(cm2/e)	Effective Diffusion Coefficient through Foundation Crecks	Eq.
Lorack	15	(cm)	Enclosed-Space Foundation or Wall Thickness	ASTM
nl	0.01	(unifices)	Areal Fraction of Cracks in Foundations/Walls	ASTM
alculated Parameter	Value	Units	Description	Source
VPwesp	6.77E-44	(mg/m3)/(mg/L)	Groundwater - Enclosed Space Vapore Volutilization Factor	Eq.
		Rick-Based Sc	reening Level for inhalation (RBSLeir)	
Input Parameters	Value	Units	Description	Source
TR	1E-06	. unitiess	Target Cencer Risk	ASTA
BW	70	(lig)	Adult Body Weight	ASTA
ATC	70	(years)	Averaging Time for Cercinogens	ASTN
Military Landanas	15	(m3/d)	Daily Indoor Inhalation Rate	ASTR
IRair-indoor	360	(d/yr)	Exposure Frequency	ASTA
EF				ASTR
ED ED	30	(yr)	Exposure Duration	
EF	30 0.4	(yr) (mg/lig-d)-1	Exposure Duration Inheletion Cencer Stope Factor	IEP/
EF ED				

S	Site Specific Parameter
ASTM .	ASTM Standard E 1739-95
TACO	Table E, Default Physical/Chemical Parameters, Part 742, June 1998
IEPA	Value provided by the IEPA
Eq	Value calculated by previous equation

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Private Residence Vicinity of Locidormer Facility Liele, Minols

Chemical Compound: Trichloroethylene

CALCULATED PARAMETERS							
Risk-Resed Screening Level for Enclosed-Space Vapor Inhelation (RESLur)							
Input Parameters	Value	Unite	Description	Source			
RBSLeir	2.84E-02	(ug/m3)	Risk-Based Screening Level for Inhalation	Eq.			
VFweep	5.77E-04	(mg/m3)/(mg/L)	Groundwater - Englosed Space Vapors Votatilization Factor	Eq.			
Calculated Parameter	Value	Units	Description	Source			
RESLW	4.82E-02	(ma/L)	Righ-Based Screening Level for Enclosed-Space Vapor Inhelation	Eq.			

S Site Specific Parameter
ASTM ASTM Standard E 1739-95

TACO Table E, Default Physical/Chemical Parameters, Part 742, June 1998

tEPA Value provided by the IEPA

Eq Value calculated by previous equation

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